DATA SHEET

S 3000

CERAMIC COLOR & CHEMICAL MFG. CO.

412-846-4000

5140

BLACK COBALT OXIDE 7:

TYPICAL COMPOSITION %

Co 72.0 November 7, 1985

HAZARDOUS INGREDIENTS

Hazardous Ingredients	Calculated Composition %	C.A.S. No.	PEL'-mg/m ³	TLV ² -mg/m ³
Cobalto-Cobaltic oxide (Co ₃ O ₄)	> 95	1308-06-1	0.1 as Co	0.1 as Co3
Cobaltous oxide (CoO)	< 5	1307-96-1	0.1 as Co	0.1 as Co3
Nickel oxide (NiO)	. 0.9	1313-99-1	1 as Ni	1 as Ni

PHYSICAL and CHEMICAL DATA

Black, odorless powder of which 98% passes through a 350 mesh sieve.

Ingredient	Molwt.	Density g/cm ³	m.p.	b.p. <u>°C</u>	Sol. in H ₂ O g/100 ml	Magnetic properties
Co ₃ O ₄	240.80	6.07	_	-	0	paramagnetic
C ₀ O	74.93	6.45	1935	-	0	paramagnetic
NiO .	74.71	6.67	1990	-	0	paramagnetic

PHYSICAL HAZARDS

None

HEALTH HAZARDS

Inhalation:

Some workers engaged long-term in the production of cobalt oxides showed symptoms of chronic bronchitis.

Inhalation experiments show that cobalto-cobaltic oxide accumulated in the lymph nodes of dogs suggesting it is tissue insoluble.

Excessive concentrations of airborne cobaltous oxide resulted in the development of emphysema and pneumoconiosis in hamsters.

The inhalation of cobaltous oxide and of nickel oxide, even at high concentrations, has not resulted in an increased incidence of malignant tumors in rodents.

The National Toxicology Program has listed nickel oxide as a possible cancer hazard. The International Agency for Research on Cancer concluded there was sufficient evidence that nickel refining was carcinogenic to humans and limited evidence that nickel and certain nickel compounds were carcinogenic to humans. IARC could not state with certainty which forms of nickel are human carcinogens but said "...metallic nickel seems less likely to be so than nickel subsulphide or nickel oxides.".

The inhalation of nickel oxide impaired long-term lung clearance in rats and high concentrations produced pneumoconiosis in hamsters.

Wounds:

Cobaltous oxide and nickel oxide have caused tumors at the site of injection in rats.

Ingestion:

The oral toxicity of cobalto-cobaltic oxide and of cobaltous oxide are low. Their oral rat LD_{50s} are $Co_3O_4 > 5000$ mg/kg and CoO = 1700 mg Co/kg. Nickel oxide also has a low oral toxicity (oral rat LD₅₀ > 5000 mg/kg).

PRECAUTIONS FOR SAFE STORAGE, HANDLING AND USE

Do not inhale powder. Keep container closed when not in use. Ventilation is normally required when handling or using this product to keep exposure to airborne cobalto-cobaltic oxide below the PEL. If ventilation alone cannot so control exposure, use NIOSH-approved respirators selected according to OSHA 29 CFR 1910.134. Maintain airborne cobalto-cobaltic oxide levels as low as possible.

SPILL, LEAK AND DISPOSAL PROCEDURE

Collect spills by wet sweeping or by vacuuming with the vacuum exhaust passing through a high efficiency particulate arresting (HEPA) filter if the exhaust is discharged into the workplace.

Wear appropriate NIOSH-approved respirators if collection and disposal of spills is likely to cause the concentration of airborne cobalto-cobaltic oxide to exceed the PEL.

Cobalt-containing waste is normally collected to recover cobalt values. Should waste disposal be deemed necessary follow EPA and local regulations.

EMERGENCY AND FIRST AID PROCEDURES

Cleanse wounds thoroughly to remove any oxide particles.

OSHA Permissible Exposure Limit.

² Threshold Limit Value of the American Conference of Governmental Industrial Hygienists.

he ACGIH has been considering since 1976 whether to reduce the TLV for cobalt metal, dust and fume to 0.05 mg Co/m³.

Describes possible health hazards of the oxide product supplied. If user operations change it to other chemical forms, whether as end products, intermediates or fugitive emissions, the possible health hazards of such forms must be determined by the user.